ALKALOIDS OF Veratrum oxysepalum. II.

N. V. Bondarenko

The combined alkaloids were obtained by the diethyl ether extraction of the roots with rhizomes of *Veratrum oxysepalum* Ture. [1]. Separation was performed by column chromatography on cellulose impregnated for formamide [2] in the chloroform system. The working up of the fractions yielded three alkaloids: with R_f 0.18 (I), 0.36 (II), and 0.47 (III) (system 1: chloroform saturated with formamide; Leningrad type M [slow] paper impregnated with formamide).

Alkaloid (I) had the formula $C_{27}H_{43}NO_2$, mp 239-241°C (ethanol), $[\alpha]_D^{22} + 18°$ (c 0.41; ethanol). The UV spectrum of (I) in concentrated sulfuric acid (0.37 mg in 10 ml) obtained 24 h after dissolution had λ_{max} 250, 323, 444, 504 nm, and its R_f value in system 1 coincided with that of rubijervine; a mixture of the compound with rubijervine gave no depression of the melting point [3]. Alkaloid (II) had the formula $C_{36}H_{51}NO_{10}$, mp 270-272°C (chloroform), $[\alpha]_D^{23} - 27.1°$ (c 0.34; chloroform). The IR spectrum of (II) (0.13 mg in 10 ml of concentrated sulfuric acid) had λ_{max} 229, 322, 411, 487 nm. IR spectrum (KBr), cm⁻¹: 1741 (ester carbonyl), 1605, 1526 (aromatic ring). In the products of its alkaline hydrolysis [4] the presence of the amino alcohol zygadenine [system 2: butan-1-ol-acetic acid-water (4:1:5)] and veratric acid [system 3: butan-1-ol-1.5 N aqueous ammonia (1:1)] was established by paper chromatography with markers. The R_f value of the alkaloid in system 1 coincided with that of veratroylzygadenine and a mixture of the compound with an authentic sample gave no depression of the melting point [5].

Alkaloid (III) had the formula $C_{37}H_{59}NO_{11}$, mp 202-204°C (benzene), $[\alpha]_D^{20} - 7^\circ$ (c 0.63; pyridine). IR spectrum (KBr), cm⁻¹: 1738 (ester carbonyl),1250 (ester/ether band). The UV spectrum of the alkaloid in concentrated sulfuric acid (0.45 mg in 10 ml) obtained 24 h after dissolution had λ_{max} 246, 315, 406, 528 nm and had no region of agreement with the spectrum taken after 1.5 h (the amino alcohol germine) [6]. In the products of alkaline hydrolysis (50 mg of the alkaloid, 2 ml of 85% methanol, and 21 mg of potassium carbonate, 20-23°C, 24 h), the presence of the amino alcohol germine (system 2) and of α -methylbutyric and α -hydroxy- α -methylbutyric acids (system 3) was established by paper chromatography with markers. Taking into account the biogenetic nature of the structure of the ester alkaloids of the false hellebore [7], the acyl residues of the acids must occupy positions 3 and 15, respectively, in the amino alcohol, which corresponds to the alkaloid germerine.

The results of analysis were confirmed by the agreement of the R_f values with those of a sample of germerine and by the absence of a depression of the melting point of a mixture.

LITERATURE CITED

1.	A. L. Shinkarenko and N. V. Bondarenko, Rast. Resur., <u>2</u> , 45 (1966).
2.	N. V. Bondarenko, A. L. Shinkarenko, and G. I. Gerashchenko, Tr. Vitebsk. Tekhnol.
	Inst. Leg. Promst., 1, 120 (1970).
3.	N. V. Bondarenko, A. L. Shinkarenko, and G. I. Gerashchenko, Khim. Prir. Soedin., 854
	(1971).
4.	W. Poethke and W. Kerstan, Pharm. Zentralh., 99, No. 7, 391 (1960).
5.	J. Tomko, J. Bendik, and S. Bauer, Planta Med., 10, 138 (1962).
6.	N. V. Bondarenko, Zh. Obshch. Khim., 37, 332 (1967).

Belorussian Institute for the Advanced Training of Physicians, Minsk. Translated from

Khimiya Prirodnykh Soedinenii, No. 2, pp. 243-244, March-April, 1983. Original article submitted October 6, 1982.

7. N. V. Bondarenko, in: A Phytochemical Study of the Flora of the Belorussian SSR and Biopharmaceutical Studies of Drugs [in Russian], Leningrad (1975), p. 84.

ALKALOIDS OF Rhinopetalum korelini

K. Samikov, R. Shakirov, and S. Yu. Yunusov

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The plant *Rhinopetalum korelini* Fisch. (family Liliaceae) has not been studied chemically [1]. We have investigated the epigeal part collected on May 10, 1974, in the village of Babadurmez, Turkmen SSR, in the period of mass fruit bearing. A mixture of bases was extracted from the plant with chloroform. From the extract, after appropriate working up, were obtained the combined ether-soluble (0.07%) and chloroform-soluble (0.08%) alkaloids, amounting to 0.15% on the weight of the dry plant.

The addition of methanol to the ether fraction yielded a mixture of two substances (0.11 g) which was passed through a column of silica gel (KSK, 150-250 mµ). Elution was carried out with chloroform-methanol (10:2), and 15 fractions (10 ml each) were collected. The treatment of fractions 6-15 with methanol gave the alkaloid (I) with mp 255-257°C (methanol). The combined chloroform alkaloids (1.14 g) were passed through a column of silica gel, and elution with chloroform-methanol (10:2 and 10:4) gave 43 fractions. After rechromatography on alumina (activity grade II), fractions 1-5 yielded an alkaloid (II) with mp 209-212°C, which was identified as solanidine (mixed melting point, R_{f}) [2]. Frac-

tions 14-19 (0.26 g) yielded the alkaloid (I). The treatment of fractions 20-30 with methanol gave the alkaloid (III) with mp 301-303°C (methanol). The chloroform-methanol (10:4) eluate deposited crystals with mp 272-274°C (methanol), $[\alpha]_{\rm D}$ + 17.7° (c 1.92; pyridine)

of an alkaloid (IV).

A study of the products of hydrolysis of (I) and (III), and also a comparison of their IR spectra and melting points with those of rhinoline and rhinolinine showed that (I) and (III) were identical with rhinoline and rhinolinine, respectively [3, 4].

Thus, solanidine, rhinoline, and rhinolinine have been isolated for the first time from *Rhinopetalum korelini*.

LITERATURE CITED

Flora of Turkmenia [in Russian], Leningrad, Vol. 1, No. 2 (1932), p. 298.
K. Samikov, R. Shakirov, and S. Yu. Yunusov, Khim. Prir. Soedin., 537 (1974).
K. Samikov, R. Shakirov, and S. Yu. Yunusov, Khim. Prir. Soedin., 815 (1978).
K. Samikov, R. Shakirov, and S. Yu. Yunusov, Khim. Prir. Soedin., 350 (1979).

Institute of the Chemistry of Plant Substances, Academy of Sciences of the Uzbek SSR, Tashkent. Translated from Khimiya Prirodnykh Soedinenii, No. 2, pp. 244-245, March-April, 1983. Original article submitted October 11, 1982.

243